

## Chemistry Practicals For Class 11 Salt Analysis

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### Chemistry Practical Class 11 Experiments and Observations ...

CBSE Class 11 Chemistry Lab Manual. (a) To study the change in pH of acetic acid (a weak acid) solution by addition of sodium acetate (b) To study the change in pH of ammonium hydroxide

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(a weak base) solution by the addition of ammonium chloride. Determine the strength of the given sodium hydroxide solution.

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### **11th Grade High School Chemistry Topics - ThoughtCo**

A blog dedicated for the love and learning of Chemistry. 28.2.11. TITRATIONS ... Okay so today in class, we learned how to use the information given to us to determine the concentration of such and such substances: ... Next we add up  $11.2 \text{ mL} + 10.9 \text{ mL} + 11.4 \text{ mL} =$  divide by 3 =  $11.16666667 \text{ mL}$  \*Note we don't have to use 8.9 because it's not a ...

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### **Chem Practical Salt Analysis class XI**

Take 2ml of fruit juice in a test tube. Add 2ml of Benedict's solution to it. Boil test tube for 2 minutes & cool. (ii) FEHLING'S TEST. Take 2ml of fruit juice in a test tube & add 2ml of each of Fehling's solution A & Fehling's solution B in it & boil.

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