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ANSWERS

By looking at the periodic table (section 6 of the IB data booklet), it can be seen that the element is lithium. $30 \text{ For CdTe, percentage by mass} = 112.41 \times 100 = 46.84\% \text{ } 112.41 + 127.60$ Overall, CdS has the highest percentage by mass of cadmium. You could also approach this question by considering the Ar of the other element in the compound.

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