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B.3.1 (Part 1) Organizing the Elements in the Periodic ...

In the mid-1800's there were 64 known elements (today there are over 116 elements). Scientists kept having difficulties organizing the information about the 64 elements. They kept trying to organize them, but found no

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pattern between them. Many chemicals have very similar colours, lustre, and conductivity of electricity and heat. Scientists were able to classify only two groups of elements ...

Organize the Elements: Periodic Table |

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elements into rows, as shown in Figure 3.

Mendeleev arranged the elements into rows in order of increasing mass so that elements with similar properties were in the same column. The final arrangement was similar to a winning arrangement in solitaire, except that the columns were organized by properties instead of suits.

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5.1 Organizing the Elements

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Elements This section explains how the elements are organized in a chart called the periodic table. It also explains what information the periodic table contains. Use Target Reading Skills Before you read, preview the red headings. In the graphic organizer below, ask a what or how question for each heading.

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Elements and the Periodic Table Organizing the Elements

Section 3-1 Science
Organizing The
Elements. Question
Answer; Mendeleev
created the first
periodic table by
arranging elements in
order: Increasing
atomic mass: How did
chemists change
Mendeleev's periodic
table in the early
1900s? They used

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atomic number instead of atomic mass to organize the elements:

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The elements in Groups 1A - 8A are called the representative elements and the elements in Groups 3 - 10 are called the transition metals. The metallic elements are shown in purple. Metals

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are solids (except for mercury), can conduct electricity and are usually malleable (can be rolled into sheets) and are ductile (can be drawn into wires).

1.2: Organization of the Elements - The Periodic Table ...

Elements and the Periodic Table
Organizing the Elements Answer Key
Description Of :
Chemistry 6.1

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A logical way to begin grouping elements together was by their chemical properties. In other words, putting elements in separate groups based on how they reacted with other elements. In 1829, a German chemist, Johann Dobereiner (1780 - 1849), placed various groups of three elements into groups called triads.

6.1: Early History of

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**the Periodic Table -
Chemistry ...**

1 3.1 The Periodic
Table (Textbook p.84)

A Table of the
Elements In addition to
his Atomic Theory, John
Dalton also devised a
system of chemical
symbols and, having
ascertained the
relative weights of
atoms, arranged them
into a table. • Other
scientists also tried to
arrange the elements
in the form of a table.

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3.1.1.

3.1 The Periodic Table A Table of the Elements

Organizing the Elements. The Periodic Table. Elements. There are 118 different kinds of atoms. Most matter is made up of two or more kinds of atoms, but some consist of just one type of atom. When matter is composed of one type of atom it is called an

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element.

Organizing the Elements

By organizing various elements with their varying properties, such as state, reactivity and valence, students will understand the correlation between element properties and organization. This lesson is based on the 5E instructional model and allows students to

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discover and develop
the model for periodic
table organization via
collaborative group
work.

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